

“Introduction to Carbon Cycle Science and pH in the Ocean”

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Earth2Class Workshops for Teachers
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The Importance of the Carbon Cycle

- Carbon is the key to Life as we know it—most biological molecules (carbohydrates, lipids, proteins, DNA and RNA) are carbon-based
- Biological energy – either photosynthetic or chemosynthetic—require carbon compounds
- Hydrocarbons have served as the most essential fuels during the past 150 years
- Coal and plant fuels also are carbon-based
- C is very reactive and rarely uncombined

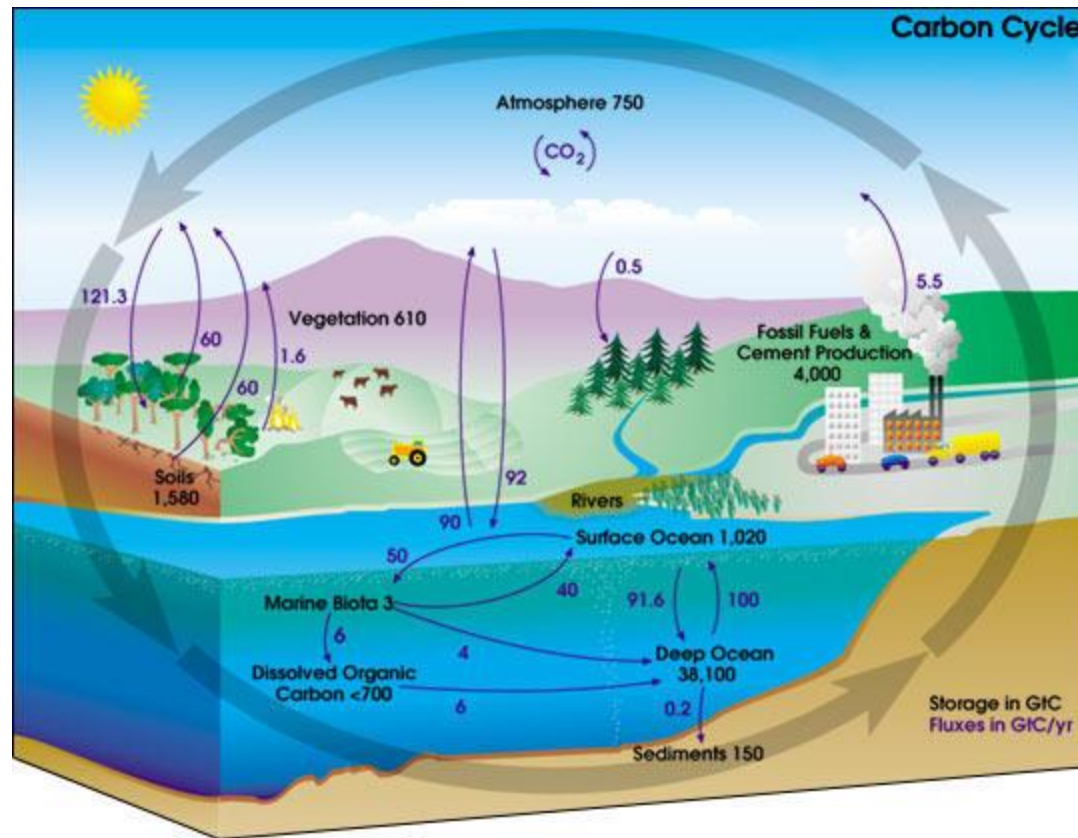
Before We Begin: Some Essential Terms

- **“Biogeochemical cycle”**: model representing movement of a substance among Earth’s 4 ‘spheres’
- **“Flux”**: amount of a quantity (such as heat or CO₂) that flows through a unit area in a unit time
- **“Reservoir”**: component of a system separate from other components, such as ‘ocean’ vs. ‘air’ or ‘land’
- **“Sequestration”**: amount of a compound ‘locked away’ in a reservoir so as not to be available

More Essential Terms

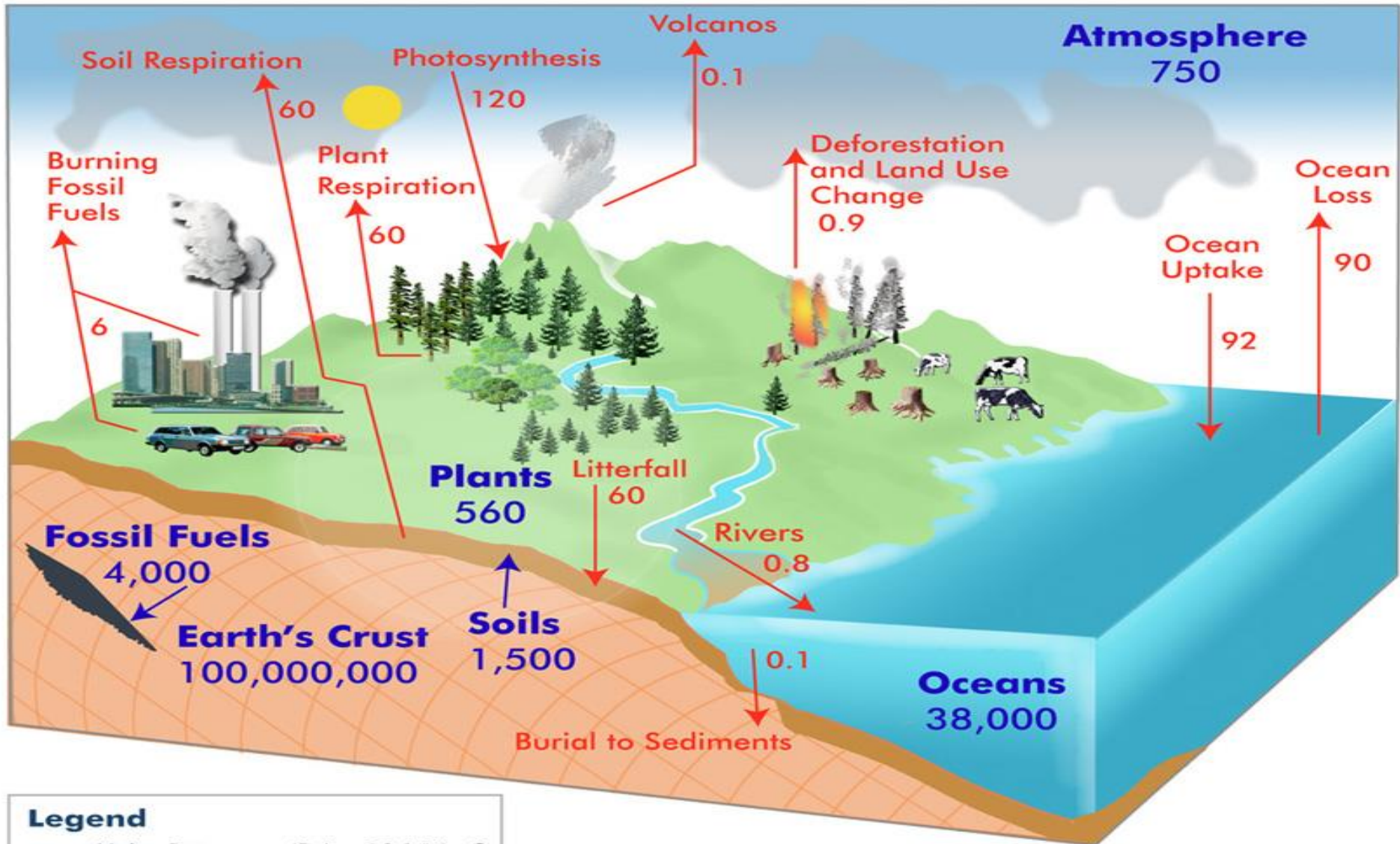
- **“Source”**: where elements or compounds come from within a system, such as the atmosphere
- **“Sink”**: where in a system substances finally wind up, such as in plants or rocks
- **“Anthropogenic CO₂”**: CO₂ released by human activities
- **“Carbonate mineral”**: compound that has “CO₃” combined with one or more elements, such as “CaCO₃” or “(Ca,Mg)CO₃”
- **“Partial Pressure”**: portion of a total amount produced by one component (e.g., pCO₂)

This drawing of the Carbon cycle indicates that plants and oceans play key roles in the flow



http://earthobservatory.nasa.gov/Library/CarbonCycle/carbon_cycle4.html

Global Carbon Cycle



Legend

Units: Petagrams (Pg) = 10^{15} gC

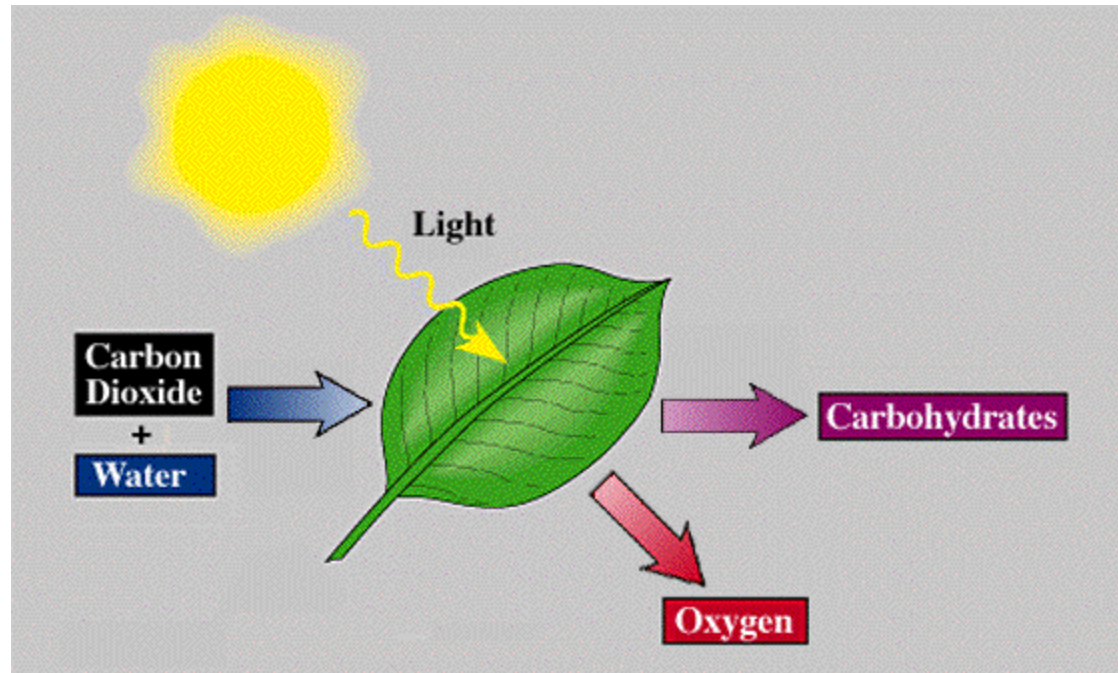
- Pools: Pg
- Fluxes: Pg/year

<http://www.globe.gov/fsl/html/templ.cgi?carboncycleDia>

Within the Carbon Cycle, the role of CO₂ has received considerable attention

- CO₂ is vital in *photosynthesis-respiration, combustion*, and other parts of the *C Cycle*
- CO₂ is an atmospheric *greenhouse gas*
- Research at the Mauna Loa observatory and elsewhere have identified sharply increased levels in the atmosphere
- CO₂ also occurs as one of the dissolved gases in seawater

Through photosynthesis, CO_2 and H_2O combine using solar energy to create more complex compounds that store energy



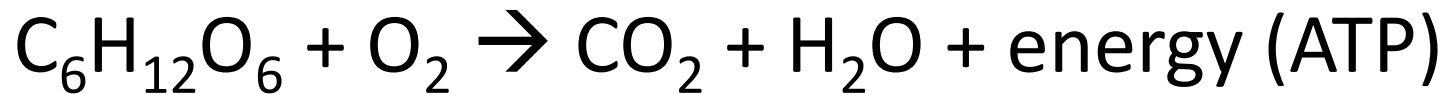
http://www.phschool.com/science/biology_place/biocoach/photosynth/overview.html

Here is another version of a photosynthesis diagram:

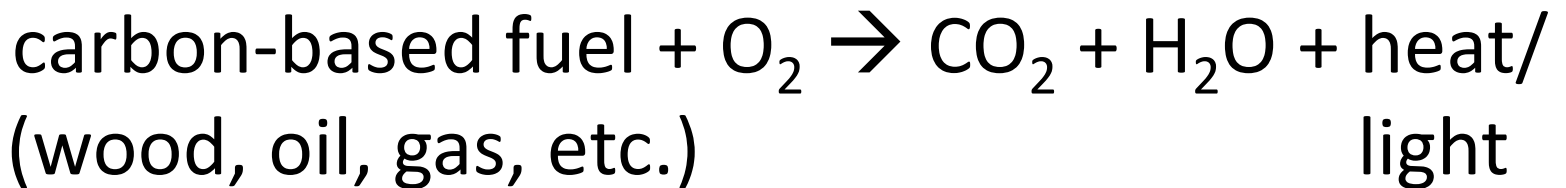
<http://earthguide.ucsd.edu/earthguide/diagrams/photosynthesis/photosynthesis.html>

Through cellular respiration, organisms gain energy and CO₂ is released from the Biosphere

In a very simplified form, **aerobic respiration** involves:



Also simplified, **combustion (burning)** involves:

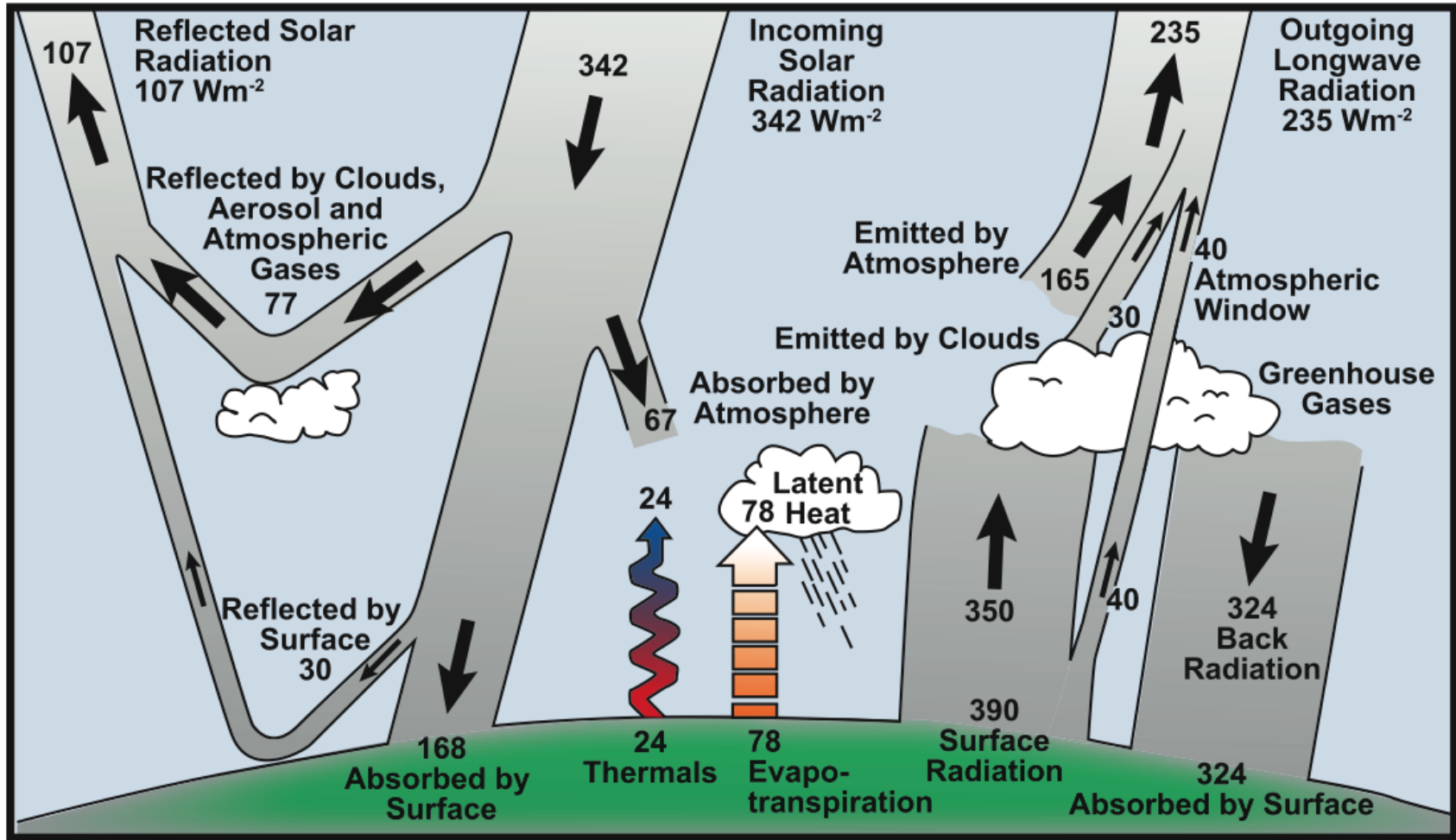


What Do We Know about CO₂ and Climate, and How Do We Know It?

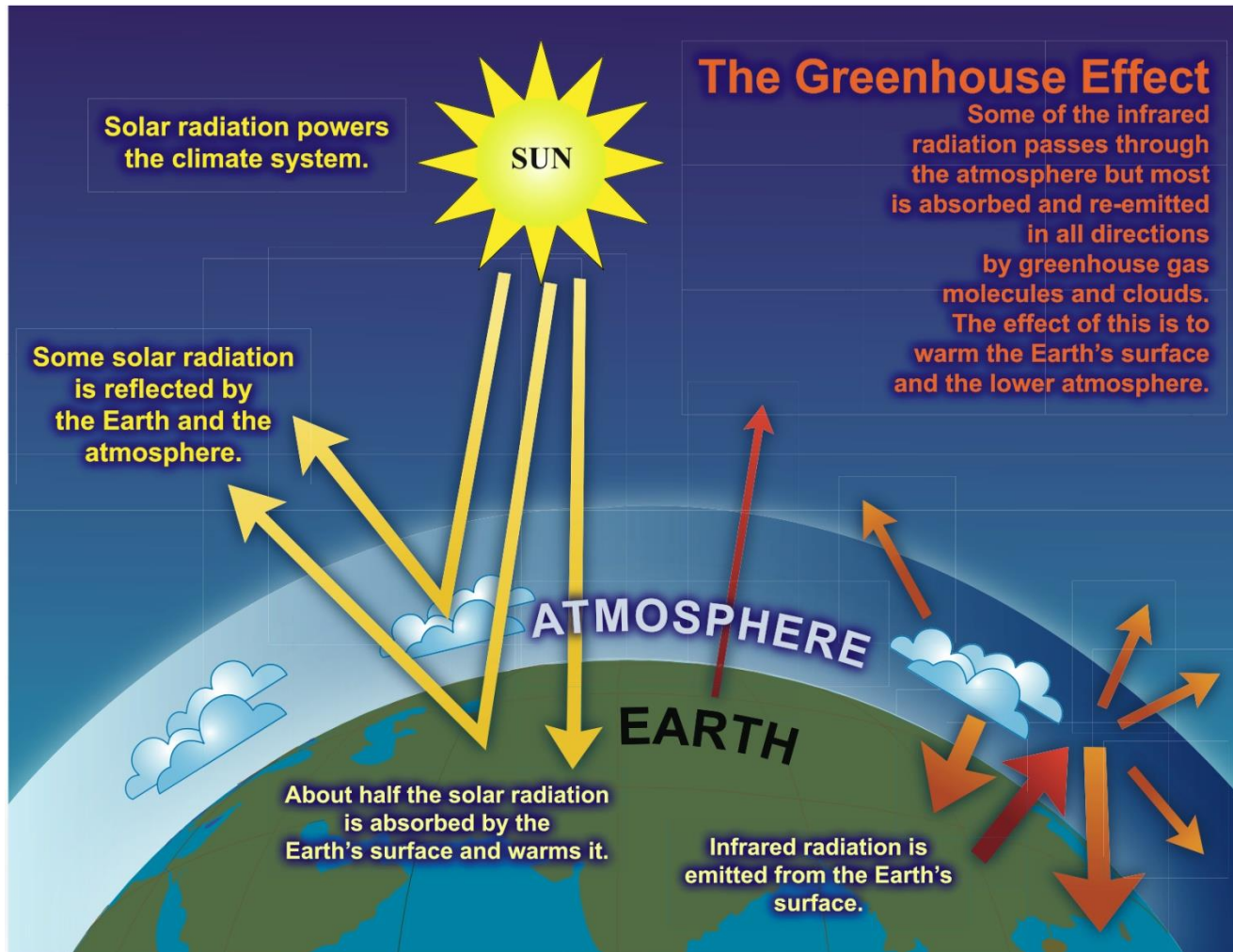
One of the better sources for information about current understanding of the changing Earth System is the [“Intergovernmental Panel on Climate Change Working Group 1: The Physical Science Basis of Climate Change”](#) report.

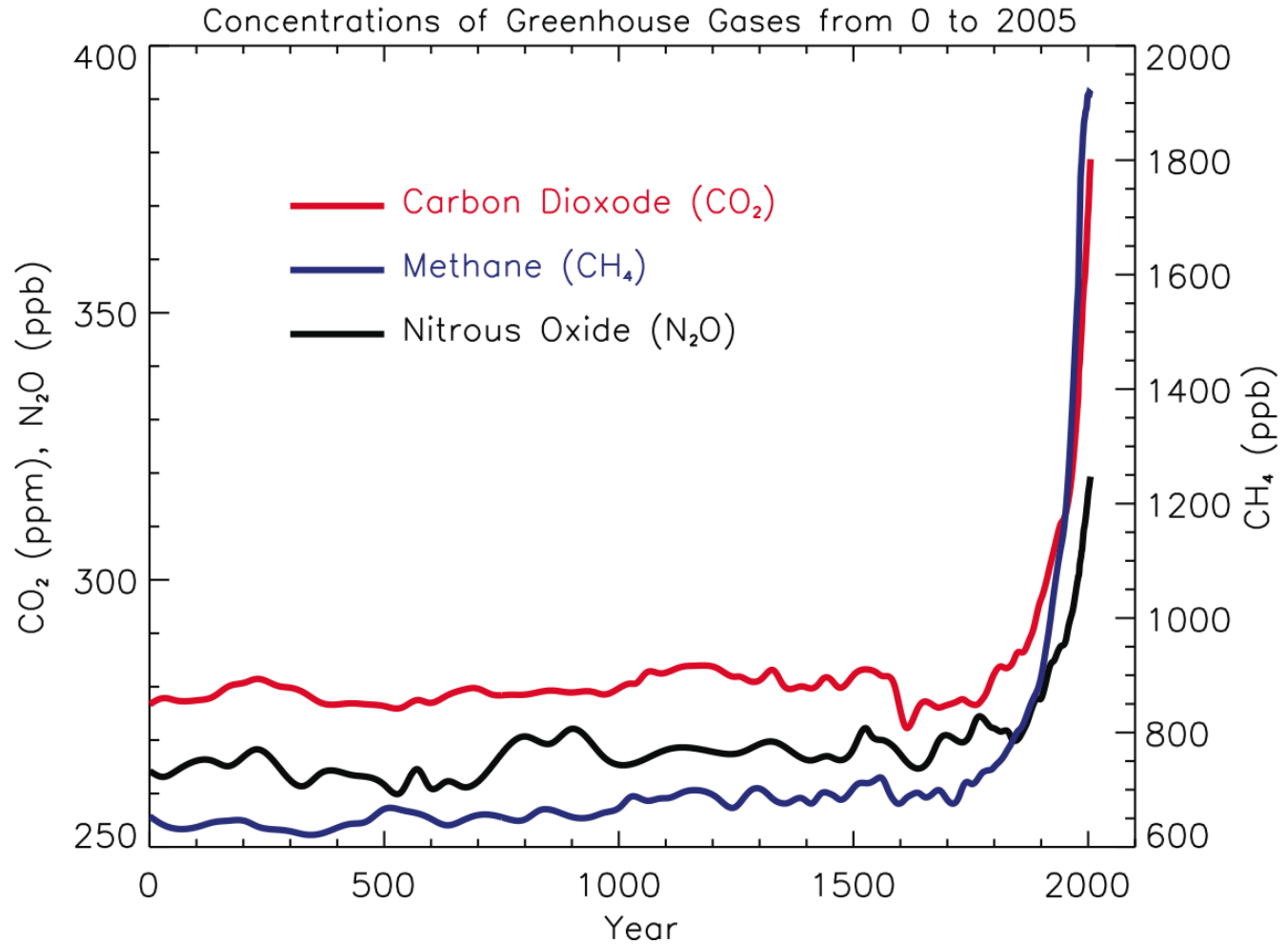
The following diagrams come from online versions of this document.

CO₂ plays a role in Earth's Energy Budget



The Greenhouse Effect





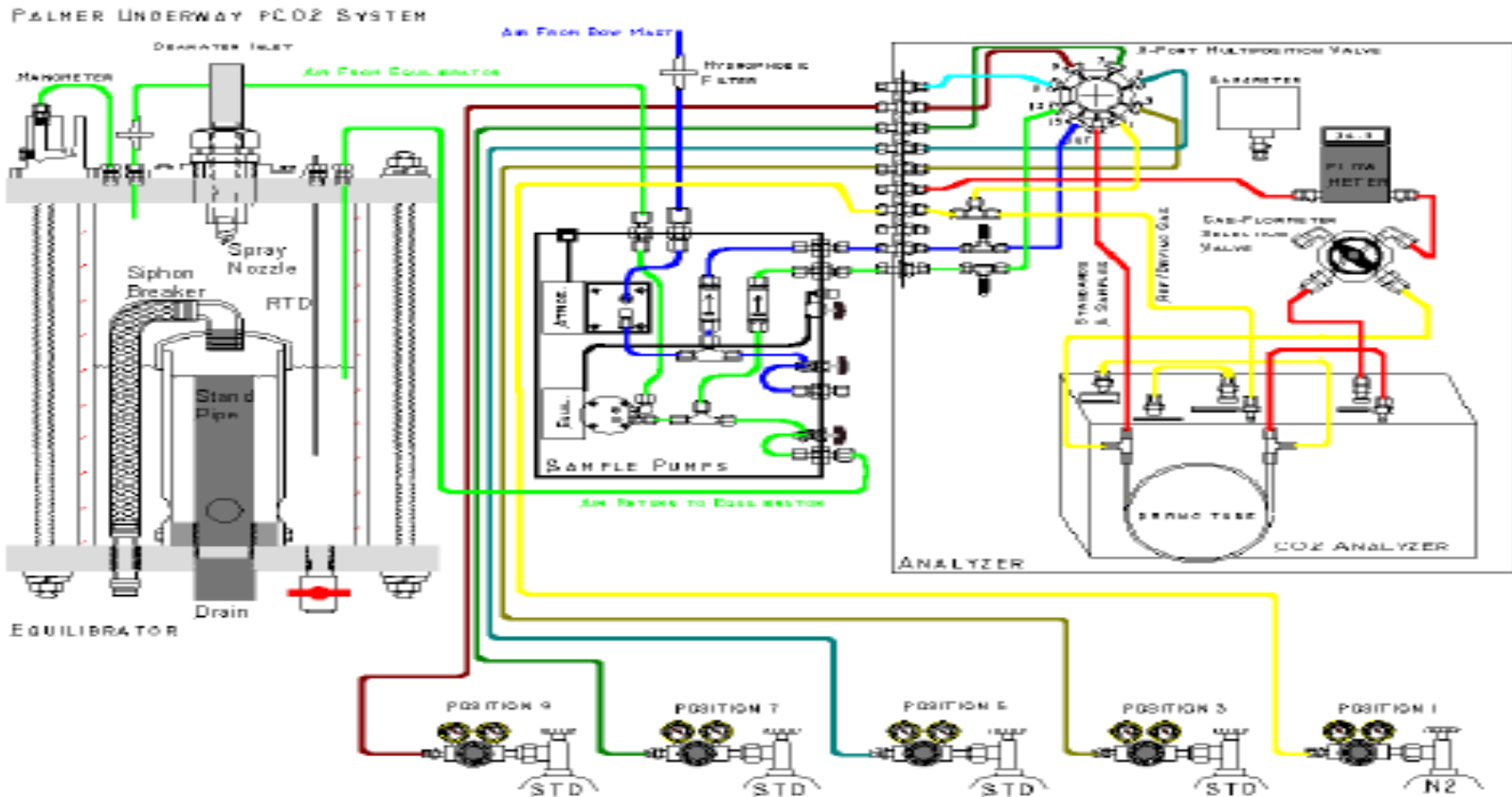
http://ipcc-wg1.ucar.edu/wg1/FAQ/wg1_faq-2.1.html

Measuring CO₂ exactly in all reservoirs and fluxes becomes critical for understanding the Earth System

- Many instruments are available to monitor CO₂ under a wide variety of conditions
- Deployment of instruments and collection of samples provide many challenges

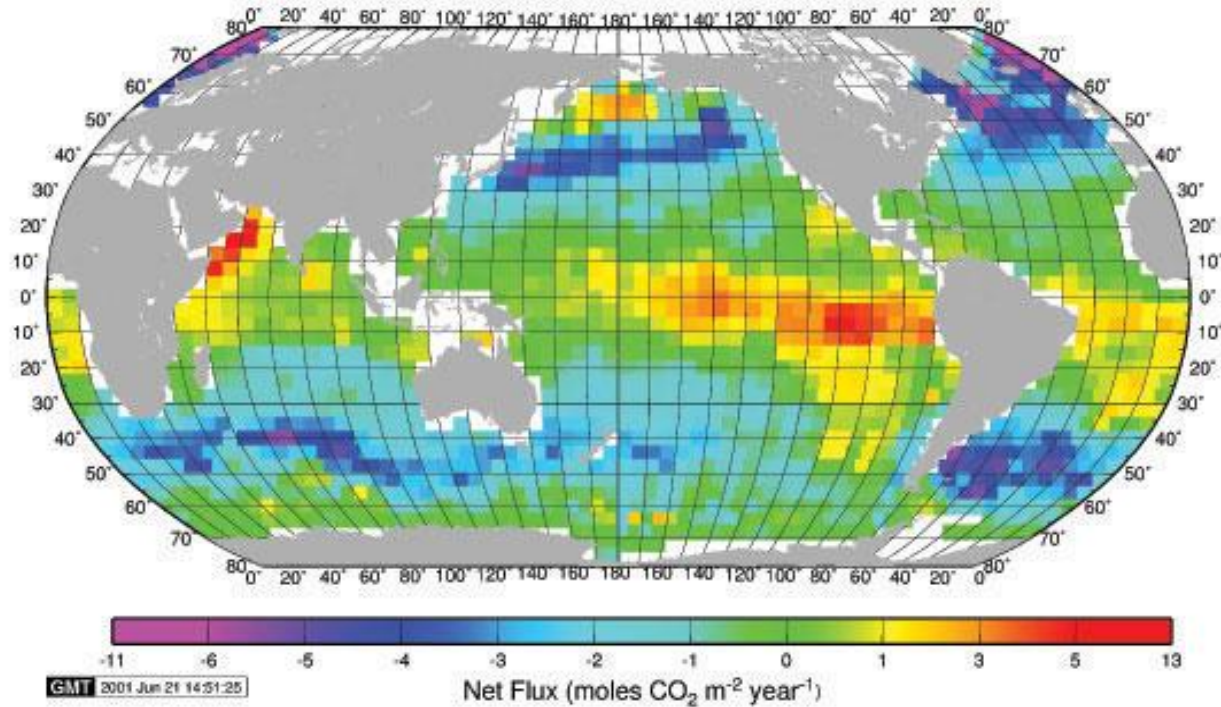


Dr. Takahashi's research focuses large on in-site measurements at sea



Data is often presented through color-coding to provides rapid visual comprehension

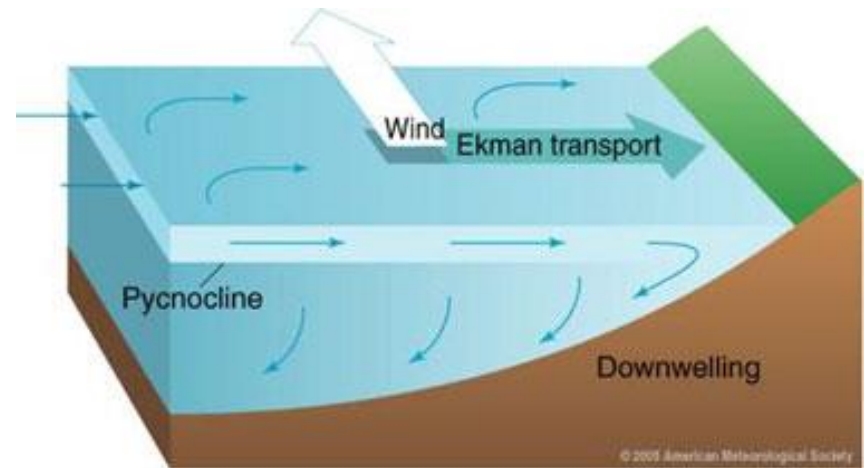
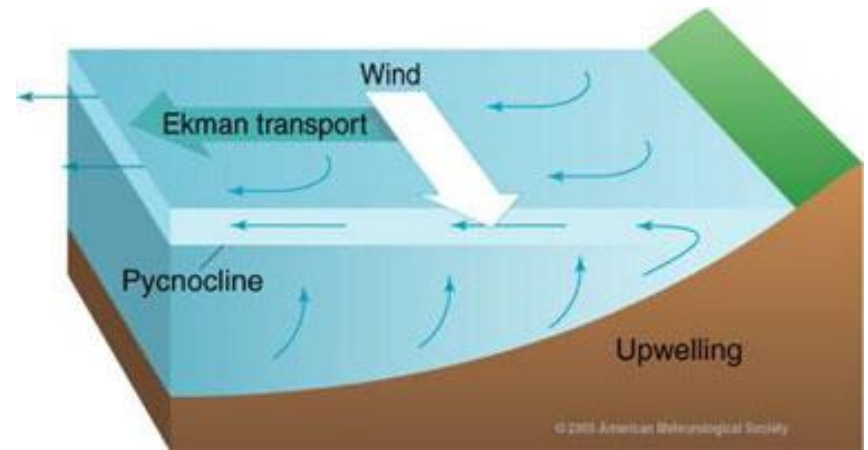
Mean Annual Air-Sea Flux for 1995 (NCEP 41-Yr Wind, 940K, W-92)



Takahashi, T. et al. (2002). Deep-Sea Res. II, 1601-1622.

In the oceans, CO₂ transport is also vertical

- Upwellings bring cooler waters (often rich in dissolved gases) to the surface
- Downwellings send surface waters into the depths and begin circulation patterns that may last for centuries



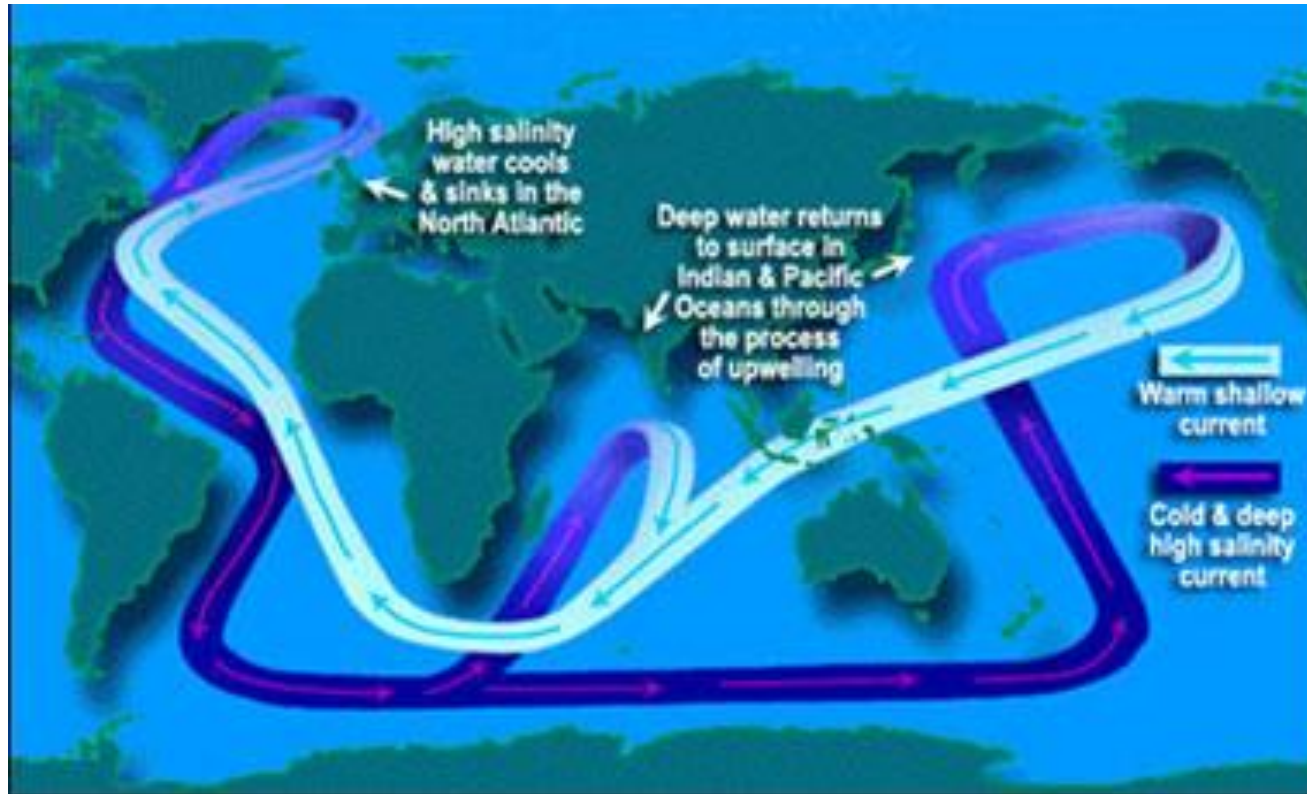
Factors Affecting Dissolving and Diffusing

- Gases dissolve in liquids.
- How much depends in part on temperature
colder = higher concentrations (think 'soda')
- Gas diffuse from higher concentrations to
lower concentrations

More about Dissolving and Diffusing

- All of this takes place amidst biogeochemical processes
- Various factors influence the rate of diffusion at the ocean-atmosphere interface
- These include surface waves and surface films
- Wind flow over the surface also adds a factor
- Marine organisms, especially phytoplankton, also influence gas exchange rates

Oceanic CO₂ flow is part of the “Ocean Conveyor Belt”



<http://oceanmotion.org/html/background/ocean-conveyor-belt.htm>

pH

- Measure of the **acidity** or **baseness (alkalinity)** of a solution
- term derived from German “power of Hydrogen”
- generally defined as the “negative logarithm of the hydrogen ion concentration”
- Acidic: less than 7 Basic: more than 7
Neutral 7.0

pH and Marine Organisms

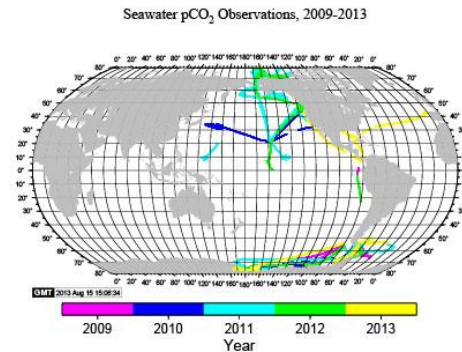
- Marine organisms live in the complex solution known collectively as “seawater”
- On the chemical scale, constant interactions among dissolved gases and solids cause a wide variation in the pH, generally from about 7.3 to more than 10
- Changes in oceanic pH can have great impacts on many marine organisms

pCO₂

- Another standard of measurement in chemical oceanography is referred to as “pCO₂”

<http://www.ldeo.columbia.edu/res/pi/CO2/carbondioxide/pages/pCO2data.html>

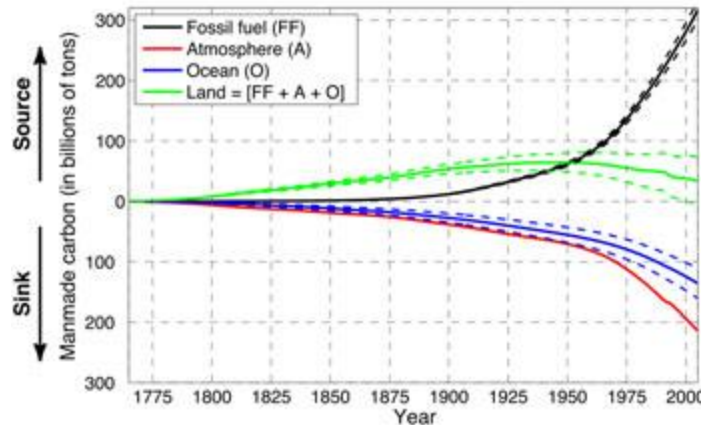
Measurements are routinely collected aboard many NOAA, Navy, Coast Guard, and other research vessels.



How Fast Are Oceans Absorbing CO₂?

- This is one of the most important questions investigated by the CO₂ group

<http://www.ideo.columbia.edu/news-events/oceans-uptake-manmade-carbon-may-be-slowing>



A Final Key Question: When Should We Begin to Teach “Climate Literacy”?

Efforts are underway to create documents clearly identifying “Essential Principles and Concepts” in all areas of the geosciences

- [“Climate Literacy: Essential Principles and Fundamental Concepts”](#)
- [Ocean Literacy](#)
- [Atmospheric Literacy](#)

So, for us as classroom teachers...

- What should we teach about CO₂ and ocean acidification?
- When should we teach it?
- How should we teach it?
- Where can we get the necessary information?
- How good are the data and the deductions?